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J.S.S ARTS, SCIENCE AND COMMERCE DEGREE COLLEGE

GOKAK



PROJECT WORK

ISOLATION AND IDENTIFICATION OF AZOTOBACTER FROM SOIL

NAME: I VITTHAL HOSUR

CLASS: B.Sc VI Sem

REG NO. S191.7247

MENTOR: Prof Dr. T.C. Gopal

Prof R M Mahendrakar



J. S. S. ARTS, SCIENCE, COMMERCE B.B.A COLLEGE AND  
P.G DEPARTMENTS, GOKAK- 591307

DEPARTMENT OF BOTANY

CERTIFICATE

This is to certify The Project report, submitted by

Mr.

VITTHAL. HOSUR Reg No. 81917247

for the Partial fulfillment of Practical in **Botany**, VI  
semester for year 2021-22 as prescribed Rani Channamma  
University Belgavi, is based on **ISOLATION AND  
IDENTIFICATION OF AZOTOBACTER FROM SOIL**  
conducted under my supervision.

HOD

Head of the  
Botany Department  
J. S. S. Science College  
GOKAK

Prof. R.M.Mahendrakar

# ISOLATION AND IDENTIFICATION OF AZOTOBACTER FROM SOIL OF Kuligod region

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## REGION

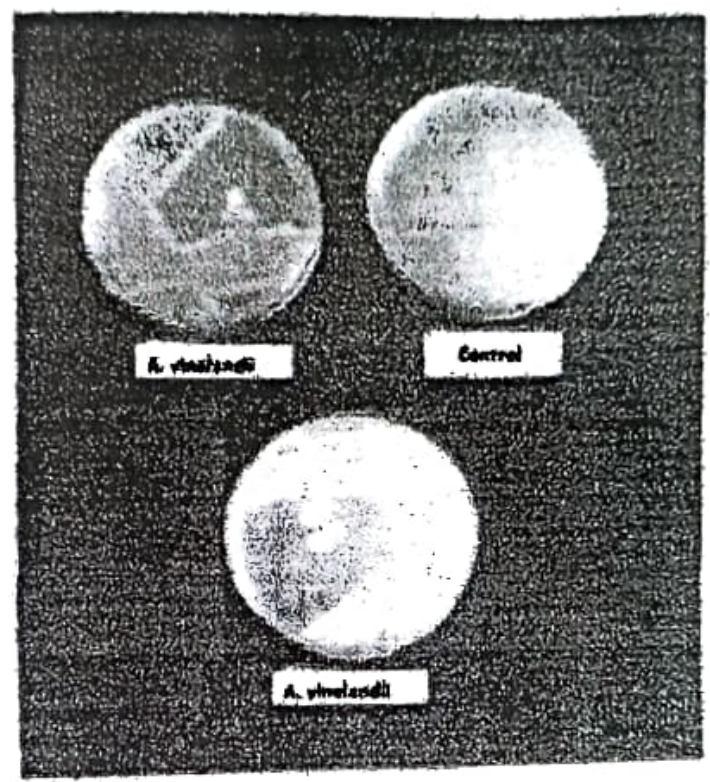
### Introduction

*Azotobacter in soil:* In Indian soils, the population of Azotobacter is not more than 10 thousand to 1 lakh/g of soil. The population of Azotobacter is mostly influenced by other microorganisms present in soil. There are some microorganisms, which stimulate the Azotobacter population in soil thereby increasing the nitrogen fixation by Azotobacter. On the other hand there are some microorganisms, which adversely affect the Azotobacter population and hence nitrogen fixation process is hampered. For example Cephalosporium is most commonly found organisms in soil, which restricts the growth of Azotobacter. Azotobacter also produces some substances, which check the plant pathogens such as Alternaria, Fusarium and Helminthosporium. Hence Azotobacter also acts as a biological control agent.

*Functions of Azotobacter:* Azotobacter naturally fixes atmospheric nitrogen in the rhizosphere. There are different strains of Azotobacter each has varied chemical, biological and other characters. However, some strains have higher nitrogen fixing ability than others. Azotobacter uses carbon for its metabolism from simple or compound substances of carbonaceous in nature. Besides carbon, Azotobacter also requires calcium for nitrogen fixation. Similarly, a medium used for growth of Azotobacter is required to have presence of organic nitrogen, micronutrients and salt in order to enhance the nitrogen fixing ability of Azotobacter.

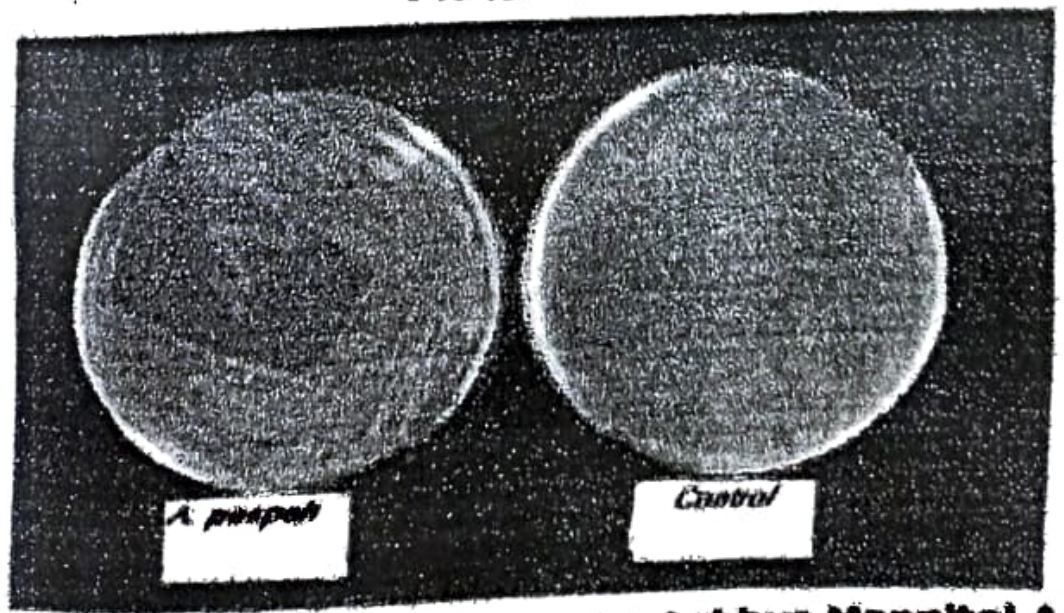
Sr. No.	Sample number	Name of the organisms
11	VMP1, VMP3,	<i>A. vinelandii</i>
21	VMP2,	<i>A. paspali</i>

**Plate No. 3.1**



**Growth of Azotobacter spp. On Ashbys Mannitol Agar**

**Plate No. 3.2**




**Growth of Azotobacter spp. On Ashbys Mannitol Agar**

## DISCUSSION

The dominant nonsymbiotic nitrogen fixing heterotrophic bacterium in Indian soils is *Azotobacter chroococcum*, *A. vinelandii*, *A. insignis*, *A. armeniacus*, *A. nigricans* and *A. beijerinckii* (Mulder and Brotonegoro, 1974). *Azotobacter* rarely exceeds  $10^4$  to  $10^5$  microorganisms per gram in Indian soils and cellulolytic microorganisms, which degrade plant residues, encourage the proliferation (Iswaran and Subba Rao, 1966).

The data showed that out of 10 soil samples collected, 3 samples were positive for non-symbiotic nitrogen fixing *Azotobacter*. On the basis of morphological, cultural and biochemical characteristics of these 3 samples showed two different species of *Azotobacter*. The morphological characterization performed as gram staining, motility, cyst formation and pigment production with the help of Bergeys manual of determinative bacteriology. Differentiation of *Azotobacter* spp. were made and identified as *Azotobacter vinelandii* and *Azotobacter paspali*

The data further indicated that, In black soil of kuligod  
*A. paspali* were less in number as compared to *A. vinelandii*

  
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**PROJECT WORK**

**ISOLATION OF RHIZOBIUM CULTURE**

**NAME :SHRUSHTI S DAPPADHULI**

**CLASS: B.Sc VI sem**

**REG.NO: S1917219**

**MENTOR: Prof.R.M.Mahindrakar**



J.S.S Degree College Gokak

### ACKNOWLEDGEMENT

We Sincerely express our deep sense of gratitude to our Prof. Mahindrakar sir, Dept. of Botany, J.S.S Degree College Gokak, for this valuable guidance, suggestions and encouragement throughout this project.

We are thankful to our HOD, Dr. T C Gopal, Dept. Of Botany, for providing necessary facilities to carryout the Project report on "ISOLATION AND IDENTIFICATION OF RHIZOBIUM FROM SOIL"

Our special thanks to our Principal Dr. A.S. Terdal for their valuable support in permitting this project report.

Date: 30/8/22  
Place: Gokak

Miss.SHRUSHTI  
Reg. No.S1917219





J .S.S ARTS, SCIENCE, COMMERCE B.B.A COLLEGE AND  
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DEPARTMENT OF BOTANY

### CERTIFICATE

This is to certify The Project report, submitted by  
Miss. SHRUSHTI S DAPPADHULI Reg.No: S1917219  
for the Partial fulfillment of Practical in Botany, VIth  
semester for year 2021-22 as prescribed Rani Channamma  
University Belgavi, is based on "ISOLATION AND  
IDENTIFICATION OF RHIZOBIUM FROM SOIL OF  
GHATAPRABHA KARNATAKA REGION" conducted under my  
supervision.

  
HOD

Dr. T.C. Gopal

Dept. of studies in Botany

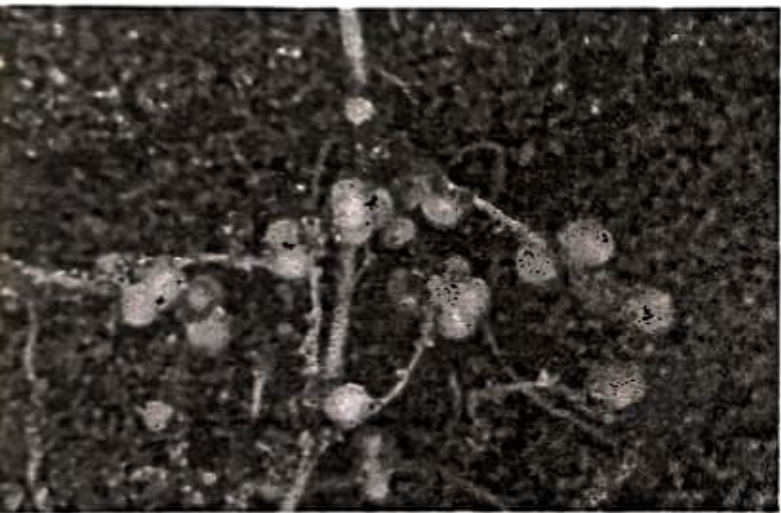
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Head of the  
Botany Department  
J. S. S. Science College  
GOKAK

  
Prof. R. M. Mahindrakar.

Dept.of studies in Botany

J.S.S Degree College





Gokak Education Society's



**J.S. S. ARTS, SCIENCE AND COMMERCE  
COLLEGE, GOKAK**

(AFFILIATED TO RANI CHANNAMMA UNIVERSITY, BELAGAVI)



**DEPARTMENT OF BOTANY**

**PROJECT WORK ON WATER ANALYSIS**

**Topic : To Determine the Alkalinity of water**

**From** ..... *Gokak falls* .....

**Class** : B.Sc Fifth Sem

**Reg No** : 21937212

**Name Of The Student** : *Smita. A. Thorat.*

  
Staff Incharge

  
HOD

## Gokak Education Society's



J.S.S ARTS, SCIENCE, COMMERCE, B.B.A COLLEGE AND P.G DEPARTMENT,  
GOKAK 591307

### DEPARTMENT OF BOTANY

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#### CERTIFICATE

THIS IS TO CERTIFY THAT THE PROJECT REPORT, SUBMITTED BY Mrs. Smita  
Appasaheb. Thorat. reg. 83917212 IN BOTANY IN 5<sup>TH</sup> SEMESTER  
2021-2022 AS PRESCRIBED BY RANI CHENNAMMA UNIVERCITY BELAGAVI IS  
BASED ON PROJECT REPORT.

HOD

*Gopal*  
**Dr. T C GOPAL**

Dept of studies in botany  
J S S Degree College, Gokak

*Valued*  
*M. J. Sir*

*M. J. Sir*  
**Prof. Mahendrakar Sir**

Dept of studies in botany  
J S S Degree College, Gokak

Determination of water analysis

Date Tested

: 22/01/2022

Tested By

: Smita A. Thorat

Project Name

: To determine the alkalinity of water

Sample Number

: 01

Sample Location

: Gokak

Sample Description

: River water, Gokak falls.

Table -2 Total Alkalinity:

Sl.No.	Volume of Sample (mL)	Burette Reading (mL)		Volume of Sulphuric acid (mL)
		Initial	Final	
1.	50	0.3	4.4	4.1
2.	50	6.0	10.2	4.2
3.	50	10.2	14.3	4.1

Calculation:

mean = 4.1

Volume of Sulphuric Acid: 4.1 mL  
 Normality of Sulphuric: 0.02 N  
 Volume of Sample: 50 mL  
 Equivalent weight of CaCO<sub>3</sub>: 1000

Total Alkalinity =  $\frac{(\text{volume of H}_2\text{SO}_4(v1) \cdot \text{Normality} \cdot 50 \cdot 1000)}{\text{Volume of sample taken}}$

Alkalinity as CaCO<sub>3</sub> equivalent (mg/L) =  $\frac{4.1}{50} \times 0.02 \times 50 \times 1000/100$   
 = 41 mg/L as CaCO<sub>3</sub> equivalent

Sample J - P = 0 T = 4.1 → T = 41

$P \times \frac{1}{2} T$  P = 0, T = 41

$\text{CO}_3^{2-} = 2P = 0$

$\text{HCO}_3^- = T - 2P = 41 - 2 \times 0 = 41 \text{ mg/L}$

Determination of water analysis Data Sheet

Date Tested : 31/01/2020  
 Tested By : Smrida A Thorat  
 Project Name : To determine the alkalinity of water  
 Sample Number : 08  
 Sample Location : Epokak  
 Sample Description :

Table -1 Phenolphthalein Alkalinity:

Sl.No.	Volume of Sample (mL)	Burette Reading (mL)		Volume of Sulphuric acid (mL)
		Initial	Final	
1.	50	16	16.3	0.3
2.	50	19	19.4	0.4
3.	50	21	21.3	0.3

Calculation:

Volume of Sulphuric Acid: 0.3 mL  
 Normality of Sulphuric: 0.02 N  
 Volume of Sample: 50 mL  
 Equivalent weight of CaCO<sub>3</sub>: 1000  
 Phenolphthalein Alkalinity =  $\frac{(\text{volume of H}_2\text{SO}_4(v1) \cdot \text{Normality} \cdot 50 \cdot 1000)}{\text{Volume of sample taken}}$

Alkalinity as CaCO<sub>3</sub> equivalent (mg/L) =  $\frac{0.3}{50} \times 0.02 \times 50 \times 1000/100$   
 = 3 mg/L as CaCO<sub>3</sub> equivalent


## INFERENCE

Alkalinity is a measure of the capacity of water to neutralize acids. The predominant chemical system present in natural waters is one where carbonates, bicarbonates and hydroxides are present. The bicarbonate ion is usually prevalent. However, the ratio of these ions is a function of pH, mineral composition, temperature and ionic strength. Water may have a low alkalinity rating but a relatively high pH or vice versa, so alkalinity alone is not of major importance as a measure of water quality. Alkalinity is not considered detrimental to humans but is generally associated with high pH values, hardness and excess dissolved solids. High alkalinity waters may also have a distinctly flat, unpleasant taste. Based on the testing, it is found that the alkalinity of the sample is  $\text{I} = 205 \text{ mg/L}$   $\text{II} = 11 \text{ mg/L}$ . As per the provisional code, alkalinity should not exceed 200 mg/L for potable water. For the fresh water alkalinity ranges between 20 – 100 mg/L. Alkalinity of tested sample is within/above the limits specified in the standards.

Hence the water sample is fit / unfit for drinking.

So,

Sample I is unfit for drinking.  
Sample II is fit for drinking.

  
Co-ordinator, IQAC  
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J.S.S ARTS, SCIENCE, COMMERCE, B.B.A. COLLEGE  
AND P,G. DEPARTMENTS, GOKAK.-591307  
**DEPARTMENT OF BOTANY**

## CERTIFICATE

*This is to certify that the study tour report, Submitted by Sachin. Rajanagol (Reg No: S1817197), for the partial fulfillment of practical in Botany, 3<sup>rd</sup> semester for year 2019-20 as prescribed by Rani Channamma University Belagavi, is based on study tour conducted under my supervision.*

HOD

*Dr. T C Gopal*

Dept. of studies in Botany

J.S.S. Degree College, Gokak

*prof. H S Dasar*

Dept. of studies in Botany

J.S.S. Degree college , Gokak

**JSS ARTS SCIENCE COMMERCE DEGREE  
COLLEGE GOKAK**



**BOTANY PROJECT**

**METHODS USED IN EFFLUENT TREATMENT BY  
INDUSTRIES**

**CLASS** : BSC V<sup>th</sup>

**SUBJECT** : Botany - II

**REG.NO** : 51716161

Prema. V. Meti

**SIGNATURE OF TEACHER**

**SIGNATURE OF H.O.D**





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J. S. S. ARTS, SCIENCE, COMMERCE, B.B.A. COLLEGE AND P.G. DEPARTMENTS, GOKAK - 591 307

## DEPARTMENT OF BOTANY

### PROJECT WORK

NAME: SHREYA.S.BANAKARI

REG : S1616223

TOPIC -  
INTEGRATED FARMING

HOD:

Dr.T.C.Gopal

Dept of studies in Botany

J.S.S Degree college Gokak

Staff:

Prof: Mahendrakar & Sudharshan

Dept of studies in Botany

J.S.S Degree college Gokak

*Examined by: Dr. T. C. Gopal*

*2)*

*Sudharshan*  
25/10/18

## Problems of present agriculture

- ❖ Decline in agriculture growth rate
- ❖ Decline in factor productivity
- ❖ Static or decline in food production
- ❖ Increasing malnutrition
- ❖ Shrinkage in net cultivable area
- ❖ Increasing environmental pollution
- ❖ Depleting ground water table
- ❖ Increasing cost of production
- ❖ Low farm income
- ❖ Problems of Farm labours due to large scale migration



## What is the solution?

**“Integrated Farming System”**



GETTING MORE OUT OF EVERY ACRE

Source: <http://planningcommission.nic.in/plans/planrel/fiveyr/1st/1planch18.html>

*[Signature]*  
Co-ordinator, IQAC  
J.S.S. Arts, Science &  
Commerce, College, Gokak.

*[Signature]*  
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**DEPARTMENTS, GOKAK**  
(NAAC Re-accreditation : "A" Grade Institution)  
Department of Botany

**PROJECT REPORT ON**

**"TREATMENT OF INDUSTRIAL EFFLUENTS IN A  
BIOREACTOR"**

**SUBMITTED BY:-**

**Savita Hosapeti**  
**REG NO. S1616194**

**UNDER THE GUIDANCE OF**

**Mr. Mahendrakar**

*Examiners.*  
1) *6/4/11/200*  
2) *4/12/18*

# CERTIFICATE

This is to certify that the project entitled, "TREATMENT OF INDUSTRIAL EFFLUENTS IN A BIOREACTOR" submitted by Miss Savita Hosapeti Reg No. S1616194 in partial fulfillments for the botany project, V Sem for year 2018-19 as prescribed by the Rani Channamma University Belagavi is based on "TREATMENT OF INDUSTRIAL EFFLUENTS"

HOD

Dr. T. C Gopal

Dept of studies in botany

J S S Degree College Gok

Prof . Mahendrakar

Dept of studies in botany


J S S Degree College Gokak


## Analytical methods for water quality parameters

BOD-The quantity of oxygen required by the microorganisms for the stabilization of the biological decomposable organic matter. BOD tests measure the molecular oxygen utilized during a specified incubation duration for the biochemical degradation of organic material and the oxygen used to oxidize inorganic material such as ferrous iron and sulfides. The most common BOD test consists of a 5 day period in which a sample is placed in an airtight bottle under controlled conditions temperature ( $20^{\circ}\text{C} \pm 1^{\circ}\text{C}$ ), keeping any light from penetrating the sample to prevent photosynthesis. The Dissolved Oxygen (DO) in the sample is measured before and after the 5 day incubation period, and BOD is then calculated as the difference between initial and final DO measurements. BOD can be considered a more "natural" test in determining the oxygen required to oxidize organic matter.

### COD

- Chemical oxygen demand test determines the oxygen required for chemical oxidation of organic component as well as the no. of inorganic component with the help of strong oxidant i.e. potassium dichromate.
- Once oxidation is complete, the excess potassium dichromate is titrated with ferrous ammonium sulfate (FAS) until all of the excess oxidizing agent has been reduced to  $\text{Cr}^{3+}$ . Typically, the oxidation-reduction indicator Ferroin is added during this titration step as well. Once all the excess dichromate has been reduced, the Ferroin indicator changes from blue-green to reddish-brown.
- COD is often preferred for daily analysis since it is inherently more reproducible, accounts for changing conditions and takes a short time to complete.

  
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